

Retraction of articles by H. Zhong *et al.*

H. Zhong,^{a*} S.-H. Duan,^a Y.-P. Hong,^a M.-L. Li,^a Y.-Q. Liu,^a C.-J. Luo,^a Q.-Y. Luo,^a S.-Z. Xiao,^a H.-L. Xie,^a Y.-P. Xu,^a X.-M. Yang,^{b,a} X.-R. Zeng^a and Q. Y. Zhong^c

^aCollege of Chemistry and Chemical Engineering, Provincial Key Laboratory of Coordination Chemistry, Jinggangshan University, Jian 343009, People's Republic of China, ^bInstitute of Applied Materials, Jiangxi University of Finance and Economics, Nanchang 330032, People's Republic of China, and ^cJian Training School, Jian 343000, People's Republic of China
Correspondence e-mail: huazhong06@126.com

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A series of 41 papers by H. Zhong *et al.* are retracted.

As a result of problems with the data sets and incorrect atom assignments, 41 papers by H. Zhong *et al.* are retracted. Full details of all the articles are given in Table 1.

Table 1

Details of articles to be retracted, in order of publication.

Title	Reference	DOI	Refcode
<i>Aquachlorobis(1,10-phenanthroline)cobalt(II) chloride thiourea solvate</i>	Zhong, Zeng, Liu & Luo (2006a)	10.1107/S1600536806041122	KERQEE
<i>cis-Dichlorobis(1,10-phenanthroline)cobalt(II)</i>	Zhong, Zeng & Luo (2006)	10.1107/S1600536806047295	MEQFOE
<i>Tris(quinolin-8-olato-κ²N,O)cobalt(III) glyoxal hemisolvate monohydrate</i>	Zhong, Zeng, Liu & Luo (2006b)	10.1107/S1600536806050240	MEQHEW
<i>(8-Quinololinol-κ²N,O)bis(8-quinolinolato-κ²N,O)nickel(II) glyoxal hemisolvate monohydrate</i>	Zhong, Zeng, Liu & Luo (2007)	10.1107/S1600536806053232	METVUD
<i>Aquachlorobis(1,10-phenanthroline)cobalt(II) chloride thioacetamide solvate</i>	Zhong, Zeng & Luo (2007)	10.1107/S1600536806053530	METQIM
<i>(8-Quinololinol-κ²N,O)-bis(8-quinolinolato-κ²N,O)zinc(II) glyoxal hemisolvate monohydrate</i>	Zhong, Zeng, Luo, Li & Xiao (2007)	10.1107/S1600536807001171	DEXTEG
<i>(Dimethylglyoxime-κ²N,N')bis(1,10-phenanthroline-κ²N,N')nickel(II) dinitrate dihydrate</i>	Zhong, Zeng, Yang, Luo & Li (2007a)	10.1107/S1600536807004102	YEYGOZ
<i>(Dimethylglyoxime-κ²N,N')bis(1,10-phenanthroline-κ²N,N')zinc(II) dinitrate dihydrate</i>	Zhong, Zeng, Yang, Luo & Li (2007b)	10.1107/S1600536807004096	YEYGUF
<i>Chloridobis(1,10-phenanthroline-κN,N')copper(I) hexahydrate</i>	Zhong, Zeng, Yang, Luo & Xiao (2007)	10.1107/S160053680700791X	HEGKOU1
<i>Tetrakis(pyridine-κN)bis(thiocyanato-κN)cobalt(II)</i>	Zhong, Zeng, Yang & Luo (2007a)	10.1107/S1600536807017461	ITCPCO1
<i>Tetrakis(pyridine-κN)bis(thiocyanato-κN)copper(II)</i>	Zhong, Zeng, Yang & Luo (2007b)	10.1107/S160053680701879X	AVUJEG02
<i>Tetrakis(nitrato-κ²O,O')bis(4-phenylpyridine-κN)cerium(IV)</i>	Zhong, Zeng, Yang & Luo (2007c)	10.1107/S1600536807018831	CICDOI
<i>Bis(4,4'-bipyridine-κ²N,N')tetrakis(nitrato-κ²O,O')cerium(IV)</i>	Zhong, Zeng, Yang & Luo (2007d)	10.1107/S1600536807021502	YIDNEF
<i>(1,10-Phenanthroline)tris(phenoxyacetato)lanthanum(III)</i>	Zhong, Zeng, Yang, Luo & Xu (2007)	10.1107/S1600536807027171	EDUROL
<i>(1,10-Phenanthroline)tris(phenoxyacetato)cerium(III)</i>	Zhong, Yang, Luo & Xu (2007a)	10.1107/S1600536807028061	EDUTUT
<i>(1,10-Phenanthroline)tri(3-phenylpropanoato)lanthanum(III)</i>	Zhong, Yang, Luo & Xu (2007b)	10.1107/S1600536807028693	RIGQEE
<i>(1,10-Phenanthroline-κ²N,N')tris(phenoxyacetato)-κO;κO;κO,O'-neodymium(III)</i>	Zhong, Yang, Luo & Xu (2007c)	10.1107/S1600536807030371	UDUMEM
<i>Bis(2,2'-bipyridyl-κ²N,N')bis(thiocyanato-κN)nickel(II)</i>	Zhong, Yang, Luo & Xu (2007d)	10.1107/S1600536807031613	YEJGOJ01
<i>Bis(2,2'-bipyridyl-κ²N,N')bis(isothiocyanato-κN)copper(II)</i>	Zhong, Yang, Luo & Xu (2007e)	10.1107/S1600536807033181	UFAPOH
<i>Bis(2,2'-bipyridyl-κ²N,N')bis(thiocyanato-κN)zinc(II)</i>	Zhong, Yang, Luo & Xu (2007f)	10.1107/S1600536807035337	TIGFAR
<i>(1,10-Phenanthroline-κ²N,N')tris(3-phenylpropanoato-κO)neodymium(III)</i>	Zhong, Yang, Luo & Xu (2007g)	10.1107/S1600536807035350	TIGFEV
<i>2-Fluoro-3,5-dinitrobenzamide monohydrate</i>	Zhong, Yang, Xie & Luo (2007j)	10.1107/S1600536807038676	VIKGAY
<i>2-Fluoro-3,5-dinitrobenzoic acid-ammonia (1/1)</i>	Zhong, Yang, Xie & Luo (2007k)	10.1107/S1600536807039724	KILKIA
<i>1-Hydroxy-4,6-dinitropyridine-2-carboxamide monohydrate</i>	Zhong, Yang, Xie & Luo (2007l)	10.1107/S1600536807040779	AFETAH
<i>N-(2-Hydroxyphenyl)carbamic acid-ammonia (1/1)</i>	Zhong, Yang, Xie & Luo (2007m)	10.1107/S160053680704086X	AFINAF
<i>catena-Poly[[bis(μ-anilinoacetato-κ²O:O')bis(μ-anilinoacetato-κ²O:O')bis(1,10-phenanthroline-κ²N,N')samarium(III)]-μ-anilinoacetato-κ²O:O']</i>	Zhong, Yang, Xie & Luo (2007a)	10.1107/S1600536807043528	PILDAQ
<i>2-Hydroxy-5-nitrobenzene-1,3-dicarboxylic acid monohydrate</i>	Zhong, Yang, Xie & Luo (2007n)	10.1107/S1600536807045199	XILWIZ
<i>catena-Poly[[tetra-μ-anilinoacetato-bis(1,10-phenanthroline)-dineodymium(III)]-di-μ-anilinoacetato]</i>	Zhong, Yang, Xie & Luo (2007b)	10.1107/S1600536807048489	WIMWEV
<i>Hexaaquacopper(II) bis(4-methylbenzenesulfonate)</i>	Zhong, Yang, Xie & Luo (2007c)	10.1107/S1600536807049525	TOLSCV01

Table 1 (continued)

Title	Reference	DOI	Refcode
<i>catena-Poly[[tetra-μ-anilinoacetato-bis(1,10-phenanthroline)-dilanthanum(III)]-di-μ-anilinoacetato]</i>	Zhong, Yang, Xie & Luo (2007d)	10.1107/S1600536807051240	GIMZEI
<i>Hexaaquachromium(II) bis(4-methylbenzenesulfonate)</i>	Zhong, Yang, Xie & Luo (2007e)	10.1107/S1600536807051227	GIMZIM
<i>Hexaaquamanganese(II) bis(4-methylbenzenesulfonate)</i>	Zhong, Yang, Xie & Luo (2007f)	10.1107/S1600536807052051	QUKQES01
<i>catena-Poly[(acetato-κO)(1,10-phenanthroline-κ^2N,N')cobalt(II)]-μ-acetato-κ^2O:O']</i>	Zhong, Yang, Xie & Luo (2007g)	10.1107/S1600536807053494	NIQLAB
<i>Hexaaquanickel(II) bis(4-aminobenzenesulfonate)</i>	Zhong, Zhong, Xie & Luo (2007a)	10.1107/S1600536807054372	HIPZOW
<i>catena-Poly[(acetato-κO)(1,10-phenanthroline-κ^2N,N')copper(II)]-μ-acetato-κ^2O:O']</i>	Zhong, Yang, Xie & Luo (2007h)	10.1107/S160053680705622X	XIRGOV
<i>Hexaaquazinc(II) bis(4-aminobenzenesulfonate)</i>	Zhong, Zhong, Xie & Luo (2007b)	10.1107/S1600536807056498	XIRJEO
<i>catena-Poly[(acetato-κO)(1,10-phenanthroline-κ^2N,N')nickel(II)]-μ-acetato-κ^2O:O']</i>	Zhong, Yang, Xie & Luo (2007i)	10.1107/S1600536807058540	HIQJOH
<i>Hexaaquacobalt(II) bis(4-aminobenzenesulfonate)</i>	Zhong, Xie & Luo (2007)	10.1107/S1600536807058527	HIQJUN
<i>catena-Poly[[tetra-μ-anilinoacetato-bis(1,10-phenanthroline)-dieuropium(III)]-di-μ-anilinoacetato]</i>	Zhong, Yang, Duan & Hong (2007)	10.1107/S1600536807060643	YIQMAN
<i>(Dimethylglyoxime-κ^2N,N')bis(1,10-phenanthroline-κ^2N,N')copper(II) dintrate dihydrate</i>	Zhong, Yang, Luo & Li (2007)	10.1107/S1600536807061193	YIQNUI
<i>catena-Poly[(1,10-phenanthroline-κ^2N,N')praseodymium(III)]-di-μ-phenoxyacetato-κ^4O:O'-[(1,10-phenanthroline-κ^2N,N')praseodymium(III)]-di-μ-phenoxyacetato-κ^4O:O'-di-μ-phenoxyacetato-κ^3O,O':κ^3O:O,O']</i>	Zhong, Yang, Luo & Xu (2008)	10.1107/S1600536807068614	GISJIC

References

- Zhong, H., Xie, H.-L. & Luo, C.-J. (2007). *Acta Cryst.* **E63**, m3054.
- Zhong, H., Yang, X.-M., Duan, S.-H. & Hong, Y.-P. (2007). *Acta Cryst.* **E63**, m3142–m3143.
- Zhong, H., Yang, X.-M., Luo, C.-J. & Li, M.-L. (2007). *Acta Cryst.* **E63**, m3160–m3161.
- Zhong, H., Yang, X.-M., Luo, Q.-Y. & Xu, Y.-P. (2007a). *Acta Cryst.* **E63**, m1885–m1886.
- Zhong, H., Yang, X.-M., Luo, Q.-Y. & Xu, Y.-P. (2007b). *Acta Cryst.* **E63**, m1909.
- Zhong, H., Yang, X.-M., Luo, Q.-Y. & Xu, Y.-P. (2007c). *Acta Cryst.* **E63**, m2019.
- Zhong, H., Yang, X.-M., Luo, Q.-Y. & Xu, Y.-P. (2007d). *Acta Cryst.* **E63**, m2062.
- Zhong, H., Yang, X.-M., Luo, Q.-Y. & Xu, Y.-P. (2007e). *Acta Cryst.* **E63**, m2141.
- Zhong, H., Yang, X.-M., Luo, Q.-Y. & Xu, Y.-P. (2007f). *Acta Cryst.* **E63**, m2208.
- Zhong, H., Yang, X.-M., Luo, Q.-Y. & Xu, Y.-P. (2007g). *Acta Cryst.* **E63**, m2209–m2210.
- Zhong, H., Yang, X.-M., Luo, Q.-Y. & Xu, Y.-P. (2008). *Acta Cryst.* **E64**, m317–m318.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007a). *Acta Cryst.* **E63**, m2508–m2509.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007b). *Acta Cryst.* **E63**, m2680–m2681.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007c). *Acta Cryst.* **E63**, m2724–m2725.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007d). *Acta Cryst.* **E63**, m2772–m2773.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007e). *Acta Cryst.* **E63**, m2774.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007f). *Acta Cryst.* **E63**, m2825.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007g). *Acta Cryst.* **E63**, m2895–m2896.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007h). *Acta Cryst.* **E63**, m2979.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007i). *Acta Cryst.* **E63**, m3053.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007j). *Acta Cryst.* **E63**, o3780.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007k). *Acta Cryst.* **E63**, o3831.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007l). *Acta Cryst.* **E63**, o3881.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007m). *Acta Cryst.* **E63**, o3882.
- Zhong, H., Yang, X.-M., Xie, H.-L. & Luo, C.-J. (2007n). *Acta Cryst.* **E63**, o4191.
- Zhong, H., Zeng, X.-R., Liu, Y.-Q. & Luo, Q.-Y. (2006a). *Acta Cryst.* **E62**, m2925–m2927.
- Zhong, H., Zeng, X.-R., Liu, Y.-Q. & Luo, Q.-Y. (2006b). *Acta Cryst.* **E62**, m3557–m3559.
- Zhong, H., Zeng, X.-R., Liu, Y.-Q. & Luo, Q.-Y. (2007). *Acta Cryst.* **E63**, m187–m189.
- Zhong, H., Zeng, X.-R. & Luo, Q.-Y. (2006). *Acta Cryst.* **E62**, m3330–m3332.
- Zhong, H., Zeng, X.-R. & Luo, Q.-Y. (2007). *Acta Cryst.* **E63**, m221–m223.
- Zhong, H., Zeng, X.-R., Luo, Q.-Y., Li, M.-L. & Xiao, S.-Z. (2007). *Acta Cryst.* **E63**, m492–m494.
- Zhong, H., Zeng, X.-R., Yang, X.-M. & Luo, Q.-Y. (2007a). *Acta Cryst.* **E63**, m1379.
- Zhong, H., Zeng, X.-R., Yang, X.-M. & Luo, Q.-Y. (2007b). *Acta Cryst.* **E63**, m1445.
- Zhong, H., Zeng, X.-R., Yang, X.-M. & Luo, Q.-Y. (2007c). *Acta Cryst.* **E63**, m1455.
- Zhong, H., Zeng, X.-R., Yang, X.-M. & Luo, Q.-Y. (2007d). *Acta Cryst.* **E63**, m1592–m1593.
- Zhong, H., Zeng, X.-R., Yang, X.-M., Luo, Q.-Y. & Li, M.-L. (2007a). *Acta Cryst.* **E63**, m639–m641.
- Zhong, H., Zeng, X.-R., Yang, X.-M., Luo, Q.-Y. & Li, M.-L. (2007b). *Acta Cryst.* **E63**, m642–m644.
- Zhong, H., Zeng, X.-R., Yang, X.-M., Luo, Q.-Y. & Xiao, S.-Z. (2007). *Acta Cryst.* **E63**, m826–m828.
- Zhong, H., Zeng, X.-R., Yang, X.-M., Luo, Q.-Y. & Xu, Y.-P. (2007). *Acta Cryst.* **E63**, m1868–m1869.
- Zhong, H., Zhong, Q. Y., Xie, H.-L. & Luo, C.-J. (2007a). *Acta Cryst.* **E63**, m2913–m2914.
- Zhong, H., Zhong, Q.-Y., Xie, H.-L. & Luo, C.-J. (2007b). *Acta Cryst.* **E63**, m2990.

(1,10-Phenanthroline)tris(phenoxyacetato)lanthanum(III)

H. Zhong,^{a*} X.-R. Zeng,^a X.-M. Yang,^b Q.-Y. Luo^a and Y.-P. Xu^a

^aCollege of Chemistry and Chemical Engineering, Provincial Key Laboratory of Coordination Chemistry, Jinggangshan University, Jian 343009, People's Republic of China, and ^bInstitute of Applied Materials, Jiangxi University of Finance and Economics, Nanchang 330032, People's Republic of China

Correspondence e-mail: huazhong06@126.com

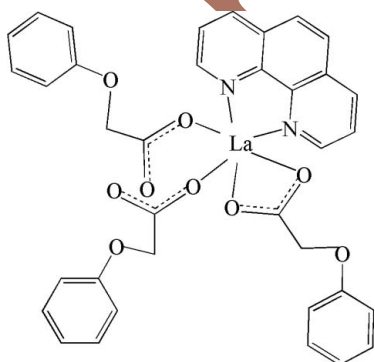
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Key indicators: single-crystal X-ray study; *T* = 273 K; mean $\sigma(\text{C}-\text{C})$ = 0.009 Å; *R* factor = 0.036; *wR* factor = 0.090; data-to-parameter ratio = 16.5.

The La^{III} atom in the title complex, [La(C₈H₇O₃)₃(C₁₂H₈N₂)], is coordinated by two N atoms of the 1,10-phenanthroline ligand and four O atoms of three phenoxyacetate ligands. This mononuclear complex is further extended into a supra-molecular network structure *via* nonclassical hydrogen bonds between CH groups of 1,10-phenanthroline and phenoxyacetate and O atoms of neighbouring phenoxyacetate ligands.

Related literature

For related literature, see: Allen *et al.* (1987); Farrugia *et al.* (2000); Kay *et al.* (1972); Ma *et al.* (1999); Mao *et al.* (1998); Starynowicz (1991, 1993); Tsukube & Shinoda (2002); Zhang *et al.* (2005); Zeng *et al.* (2000).



Experimental

Crystal data

[La(C₈H₇O₃)₃(C₁₂H₈N₂)]
M_r = 772.52
 Monoclinic, *P*2₁/*n*
a = 20.182 (3) Å
b = 8.5307 (11) Å
c = 20.833 (3) Å
 β = 107.779 (2)°
V = 3415.5 (8) Å³
Z = 4
 Mo *K* α radiation

μ = 1.31 mm⁻¹
T = 273 (2) K
 0.33 × 0.12 × 0.08 mm

Data collection

Bruker APEXII area-detector diffractometer
 Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
*T*_{min} = 0.674, *T*_{max} = 0.903
 24692 measured reflections
 6634 independent reflections
 4671 reflections with *I* > 2 σ (*I*)
*R*_{int} = 0.040

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.036
wR(*F*²) = 0.090
S = 1.07
 6634 reflections
 403 parameters
 3 restraints
 H-atom parameters constrained
 $\Delta\rho_{\text{max}}$ = 1.43 e Å⁻³
 $\Delta\rho_{\text{min}}$ = -0.59 e Å⁻³

Table 1
 Selected bond lengths (Å).

La1—O1	2.501 (3)	La1—O8	2.826 (3)
La1—O4	2.417 (3)	La1—N1	2.750 (3)
La1—O7	2.564 (2)	La1—N2	2.707 (3)

Table 2
 Hydrogen-bond geometry (Å, °).

<i>D</i> —H... <i>A</i>	<i>D</i> —H	H... <i>A</i>	<i>D</i> ... <i>A</i>	<i>D</i> —H... <i>A</i>
C10—H10...O7 ⁱ	0.93	2.33	3.196 (5)	156
C30—H30B...O4 ⁱⁱ	0.97	2.39	3.339 (5)	166
C12—H12...O5 ⁱⁱⁱ	0.93	2.45	3.039 (5)	122
C1—H1...O2 ^{iv}	0.93	2.45	3.132 (5)	130

Symmetry codes: (i) $-x + \frac{3}{2}, y + \frac{1}{2}, -z + \frac{3}{2}$; (ii) *x*, *y* - 1, *z*; (iii) $-x + 2, -y + 1, -z + 2$; (iv) $-x + 2, -y, -z + 2$.

Data collection: APEX2 (Bruker, 2004); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Siemens, 1996); software used to prepare material for publication: SHELXTL.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BQ2017).

References

Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
 Bruker (2004). APEX2. Bruker AXS Inc., Madison, Wisconsin, USA.
 Farrugia, L. J., Peacock, R. D. & Stewart, B. (2000). *Acta Cryst.* **C56**, e435–e436.
 Kay, J., Moore, J. W. & Glick, M. D. (1972). *Inorg. Chem.* **11**, 2818–2827.
 Ma, L., Evans, O. R., Foxman, B. M. & Lin, W. B. (1999). *Inorg. Chem.* **38**, 5837–5840.
 Mao, J. G., Zhang, H. J., Ni, J. Z., Wang, S. B. & Mak, T. C. W. (1998). *J. Chem. Crystallogr.* **17**, 3999–4009.
 Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
 Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.

Siemens (1996). *SAINT* and *SHELXTL*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.
Starynowicz, P. (1991). *Acta Cryst.* **C47**, 294–297.
Starynowicz, P. (1993). *Acta Cryst.* **C49**, 1895–1897.
Tsukube, H. & Shinoda, S. (2002). *Chem. Rev.* **102**, 2389–2404.

Zeng, X.-R., Xu, Y., Xiong, R.-G., Zhang, L.-J. & You, X.-Z. (2000). *Acta Cryst.* **C56**, e325–e326.
Zhang, Z.-H., Shen, Z.-L., Okamura, T.-A., Zhu, H.-F., Sun, W.-Y. & Ueyama, N. (2005). *Cryst. Growth Des.* **5**, 1191–1197.

Article retracted

supplementary materials

Article retracted

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(1,10-Phenanthroline)tris(phenoxyacetato)lanthanum(III)

H. Zhong, X.-R. Zeng, X.-M. Yang, Q.-Y. Luo and Y.-P. Xu

Comment

In recent years, there has been great interest in the synthesis of metal organic frameworks (MOFs) with organic ligands and rare earth metals because of their novel structures, fascinating properties and important roles in special materials having optical, electronic, magnetic and biological importance potential applications (Deborah *et al.*, 2000; Farrugia *et al.*, 2000; Tsukube & Shinoda, 2002; Zhang *et al.*, 2005). These compounds are usually prepared by the reaction of rare-earth metal ions with bi- or multidentate ligands (Starynowicz, 1991, 1993; Kay *et al.*, 1972; Ma *et al.*, 1999; Zeng *et al.*, 2000; Mao *et al.*, 1998).

In the molecule of (I) (Fig. 1), the ligand bond lengths and angles are within normal ranges (Allen *et al.*, 1987). The six-coordinate environment of the La atom is completed by the two N atoms of 1,10-phenanthroline ligand and four O atoms of three phenoxyacetic acid ligands (Table 1). The La—O bond lengths are in the range 2.4175 (19) to 2.827 (2) Å. The La—N bond lengths are in the range 2.707 (2) to 2.750 (3) Å. C—H...O non-classical hydrogen bonds between C—H groups of 1,10-phenanthroline or phenoxyacetic acid and O atoms of neighbouring phenoxyacetic acid molecules, with an average C...O distances of 3.179 (9) Å, generate a layered hydrogen-bonded network (Fig. 2 and Table 2). The non-classical hydrogen-bonding interactions link the mononuclear complex into a supramolecular network structure.

Experimental

Crystals of the title compound were synthesized using hydrothermal method in a 23 ml Teflon-lined Parr bomb, which was then sealed. Lanthanum (III) chloride hexahydrate (70.7 mg, 0.2 mmol), phen (39.6 mg, 0.2 mmol), phenoxyacetic acid (91.3 mg, 0.6 mmol) and distilled water (4 g) were placed into the bomb and sealed. The bomb was then heated under autogenous pressure up to 413 K over the course of 7 d and allowed to cool at room temperature for 24 h. Upon opening the bomb, a clear colorless solution was decanted from small colorless crystals. These crystals were washed with distilled water followed by ethanol, and allowed to air-dry at room temperature.

Refinement

H atoms were positioned geometrically, with C—H = 0.93 – 0.97 Å, and constrained to ride on their parent atoms, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$. FLAT and EADP restraints of *SHELXS* applied for C31/C36.

Figures

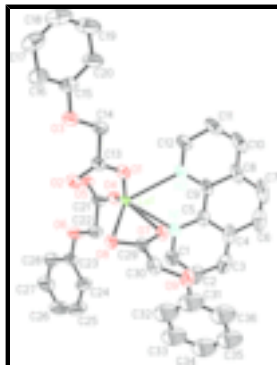


Fig. 1. View of the molecule of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level. All H atoms have been omitted for clarity.

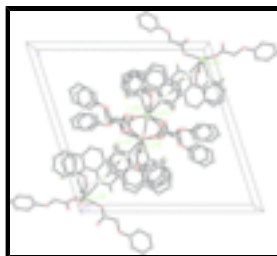


Fig. 2. A packing diagram of (I) viewed down to *b* axis. Hydrogen bonds are shown as dashed lines. Part H atoms have been omitted for clarity. [Symmetry code: (A) $2 - x, -y, 2 - z$; (B) $1 - x, 1 - y, 1 - z$; (C) $x - 1, 1/2 - y, z - 1/2$; (D) $1 - x, 1/2 + y, 3/2 - z$; (E) x, y, z ; (F) $2 - x, 1 - y, 2 - z$]

(1,10-Phenanthroline)tris(phenoxyacetato)lanthanum(III)

Crystal data

[La(C₈H₇O₃)₃(C₁₂H₈N₂)]

M_r = 772.52

Monoclinic, *P*2₁/*n*

Hall symbol: -*P* 2₁/*n*

a = 20.182 (3) Å

b = 8.5307 (11) Å

c = 20.833 (3) Å

β = 107.779 (2)°

V = 3415.5 (8) Å³

Z = 4

*F*₀₀₀ = 1552

D_x = 1.502 Mg m⁻³

Mo *K*α radiation

λ = 0.71073 Å

Cell parameters from 9081 reflections

θ = 2.5–27.0°

μ = 1.31 mm⁻¹

T = 273 (2) K

Plane, colorless

0.33 × 0.12 × 0.08 mm

Data collection

Bruker APEX II area-detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

T = 273(2) K

φ and ω scans

Absorption correction: multi-scan (SADABS; Sheldrick, 1996)

6634 independent reflections

4671 reflections with *I* > 2σ(*I*)

*R*_{int} = 0.040

θ_{max} = 26.0°

θ_{min} = 2.1°

h = -24→24

$T_{\min} = 0.674$, $T_{\max} = 0.903$
24692 measured reflections

$k = -10 \rightarrow 10$
 $l = -25 \rightarrow 25$

Refinement

Refinement on F^2	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.036$	H-atom parameters constrained
$wR(F^2) = 0.090$	$w = 1/[\sigma^2(F_o^2) + (0.0436P)^2 + 0.615P]$
$S = 1.07$	where $P = (F_o^2 + 2F_c^2)/3$
6634 reflections	$(\Delta/\sigma)_{\max} = 0.001$
403 parameters	$\Delta\rho_{\max} = 1.43 \text{ e } \text{\AA}^{-3}$
3 restraints	$\Delta\rho_{\min} = -0.59 \text{ e } \text{\AA}^{-3}$
Primary atom site location: structure-invariant direct methods	Extinction correction: none

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
La1	0.931205 (11)	0.19064 (2)	0.964291 (10)	0.03933 (9)
O1	0.97062 (14)	0.0601 (3)	0.87477 (12)	0.0530 (7)
O2	1.06694 (14)	-0.0829 (3)	0.92161 (13)	0.0557 (7)
O3	1.11837 (16)	-0.0162 (4)	0.82201 (15)	0.0735 (9)
O4	0.94007 (14)	0.4209 (3)	1.03406 (12)	0.0535 (7)
O5	1.00792 (16)	0.6128 (3)	1.09095 (13)	0.0635 (8)
O6	0.98698 (16)	0.5279 (3)	1.20623 (13)	0.0681 (9)
O7	0.84638 (13)	-0.0360 (3)	0.91880 (13)	0.0505 (7)
O8	0.94658 (13)	-0.1382 (4)	0.97709 (13)	0.0555 (7)
O9	0.78443 (16)	-0.3265 (3)	0.89809 (16)	0.0696 (9)
N1	0.79701 (17)	0.2347 (4)	0.96669 (17)	0.0533 (8)
N2	0.83407 (18)	0.3069 (4)	0.85438 (16)	0.0535 (8)
C1	0.7780 (2)	0.1918 (5)	1.0201 (2)	0.0688 (13)
H1	0.8127	0.1685	1.0600	0.083*
C2	0.7092 (3)	0.1804 (7)	1.0191 (3)	0.0892 (17)

supplementary materials

H2	0.6984	0.1506	1.0577	0.107*
C3	0.6582 (3)	0.2128 (7)	0.9617 (4)	0.099 (2)
H3	0.6119	0.2044	0.9605	0.119*
C4	0.6745 (3)	0.2591 (7)	0.9041 (3)	0.0810 (16)
C5	0.7460 (2)	0.2676 (5)	0.9087 (2)	0.0573 (11)
C6	0.6233 (3)	0.2982 (8)	0.8400 (4)	0.116 (2)
H6	0.5762	0.2901	0.8357	0.139*
C7	0.6424 (3)	0.3453 (8)	0.7875 (3)	0.109 (2)
H7	0.6084	0.3728	0.7478	0.131*
C8	0.7134 (3)	0.3547 (6)	0.7905 (2)	0.0764 (15)
C9	0.7659 (2)	0.3100 (5)	0.8504 (2)	0.0571 (11)
C10	0.7354 (4)	0.4029 (7)	0.7366 (3)	0.0936 (19)
H10	0.7028	0.4371	0.6971	0.112*
C11	0.8036 (3)	0.4011 (6)	0.7406 (2)	0.0872 (17)
H11	0.8179	0.4335	0.7043	0.105*
C12	0.8526 (3)	0.3489 (5)	0.8009 (2)	0.0692 (13)
H12	0.8993	0.3438	0.8033	0.083*
C13	1.0287 (2)	0.0035 (5)	0.87698 (19)	0.0473 (9)
C14	1.0525 (2)	0.0507 (5)	0.8173 (2)	0.0608 (11)
H14A	1.0185	0.0161	0.7759	0.073*
H14B	1.0557	0.1640	0.8157	0.073*
C15	1.1456 (3)	0.0146 (6)	0.7704 (2)	0.0679 (13)
C16	1.2080 (3)	-0.0552 (7)	0.7767 (3)	0.0873 (16)
H16	1.2293	-0.1162	0.8144	0.105*
C17	1.2400 (3)	-0.0352 (8)	0.7268 (4)	0.1020 (19)
H17	1.2824	-0.0835	0.7309	0.122*
C18	1.2094 (4)	0.0543 (8)	0.6722 (4)	0.110 (2)
H18	1.2307	0.0669	0.6387	0.132*
C19	1.1481 (4)	0.1252 (8)	0.6664 (3)	0.113 (2)
H19	1.1277	0.1876	0.6289	0.136*
C20	1.1149 (3)	0.1063 (7)	0.7158 (3)	0.0902 (17)
H20	1.0726	0.1553	0.7114	0.108*
C21	0.9692 (2)	0.4993 (5)	1.08630 (19)	0.0490 (10)
C22	0.9490 (2)	0.4462 (5)	1.14728 (19)	0.0618 (11)
H22A	0.8996	0.4638	1.1391	0.074*
H22B	0.9577	0.3346	1.1540	0.074*
C23	0.9748 (2)	0.4889 (5)	1.2657 (2)	0.0598 (11)
C24	0.9277 (3)	0.3797 (7)	1.2719 (2)	0.0801 (15)
H24	0.9014	0.3240	1.2345	0.096*
C25	0.9197 (4)	0.3526 (9)	1.3349 (3)	0.113 (2)
H25	0.8883	0.2768	1.3394	0.136*
C26	0.9566 (4)	0.4343 (9)	1.3897 (3)	0.110 (2)
H26	0.9496	0.4173	1.4312	0.132*
C27	1.0043 (3)	0.5416 (7)	1.3837 (2)	0.0969 (18)
H27	1.0305	0.5964	1.4215	0.116*
C28	1.0142 (3)	0.5702 (6)	1.3219 (2)	0.0778 (14)
H28	1.0471	0.6432	1.3181	0.093*
C29	0.8843 (2)	-0.1515 (5)	0.93930 (19)	0.0489 (10)
C30	0.8575 (2)	-0.3151 (5)	0.9191 (2)	0.0610 (11)

H30A	0.8745	-0.3503	0.8827	0.073*
H30B	0.8763	-0.3849	0.9571	0.073*
C31	0.7516 (5)	-0.3130 (8)	0.9450 (5)	0.1338 (13)
C32	0.7850 (5)	-0.2767 (8)	1.0120 (5)	0.1338 (13)
H32	0.8329	-0.2613	1.0275	0.161*
C33	0.7437 (4)	-0.2637 (8)	1.0562 (5)	0.1338 (13)
H33	0.7638	-0.2385	1.1015	0.161*
C34	0.6746 (5)	-0.2888 (8)	1.0310 (5)	0.1338 (13)
H34	0.6477	-0.2755	1.0598	0.161*
C35	0.6406 (5)	-0.3336 (8)	0.9643 (5)	0.1338 (13)
H35	0.5936	-0.3594	0.9501	0.161*
C36	0.6801 (4)	-0.3378 (8)	0.9205 (4)	0.1338 (13)
H36	0.6589	-0.3572	0.8749	0.161*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
La1	0.04562 (13)	0.03362 (13)	0.03345 (12)	-0.00164 (10)	0.00420 (8)	0.00052 (10)
O1	0.0624 (17)	0.0522 (17)	0.0435 (15)	0.0088 (15)	0.0148 (13)	0.0025 (13)
O2	0.0667 (17)	0.0541 (18)	0.0462 (16)	0.0083 (15)	0.0173 (13)	0.0111 (14)
O3	0.074 (2)	0.091 (2)	0.0632 (19)	0.0191 (18)	0.0323 (16)	0.0229 (17)
O4	0.0757 (18)	0.0386 (16)	0.0436 (15)	-0.0067 (14)	0.0140 (13)	-0.0063 (13)
O5	0.092 (2)	0.0529 (19)	0.0445 (16)	-0.0291 (17)	0.0192 (15)	-0.0019 (13)
O6	0.096 (2)	0.066 (2)	0.0420 (16)	-0.0290 (17)	0.0206 (15)	-0.0074 (14)
O7	0.0510 (15)	0.0388 (16)	0.0524 (16)	0.0003 (13)	0.0021 (12)	-0.0032 (12)
O8	0.0492 (16)	0.0594 (18)	0.0477 (16)	-0.0039 (13)	-0.0005 (13)	0.0073 (13)
O9	0.065 (2)	0.050 (2)	0.078 (2)	-0.0146 (15)	-0.0026 (16)	-0.0072 (16)
N1	0.053 (2)	0.047 (2)	0.051 (2)	0.0039 (16)	0.0044 (17)	-0.0078 (16)
N2	0.067 (2)	0.040 (2)	0.0426 (19)	0.0027 (17)	0.0014 (16)	0.0006 (15)
C1	0.063 (3)	0.083 (4)	0.059 (3)	0.008 (3)	0.017 (2)	-0.002 (3)
C2	0.061 (3)	0.126 (5)	0.085 (4)	0.005 (3)	0.030 (3)	-0.015 (3)
C3	0.054 (3)	0.120 (6)	0.119 (5)	-0.003 (3)	0.020 (3)	-0.019 (4)
C4	0.053 (3)	0.091 (4)	0.082 (4)	0.008 (3)	-0.005 (3)	-0.021 (3)
C5	0.054 (3)	0.043 (3)	0.065 (3)	0.0060 (19)	0.003 (2)	-0.011 (2)
C6	0.063 (4)	0.147 (7)	0.111 (5)	0.014 (4)	-0.014 (4)	-0.028 (5)
C7	0.081 (4)	0.120 (6)	0.089 (5)	0.038 (4)	-0.030 (3)	-0.015 (4)
C8	0.081 (4)	0.066 (3)	0.057 (3)	0.018 (3)	-0.016 (3)	-0.009 (2)
C9	0.062 (3)	0.040 (2)	0.051 (2)	0.005 (2)	-0.011 (2)	-0.005 (2)
C10	0.119 (5)	0.076 (4)	0.053 (3)	0.019 (4)	-0.022 (3)	0.001 (3)
C11	0.125 (5)	0.076 (4)	0.044 (3)	-0.002 (4)	0.001 (3)	0.011 (3)
C12	0.090 (3)	0.058 (3)	0.050 (3)	-0.003 (2)	0.006 (2)	0.008 (2)
C13	0.061 (3)	0.037 (2)	0.042 (2)	-0.004 (2)	0.0124 (19)	-0.0016 (17)
C14	0.070 (3)	0.061 (3)	0.054 (3)	0.007 (2)	0.022 (2)	0.012 (2)
C15	0.076 (3)	0.073 (3)	0.065 (3)	-0.006 (3)	0.037 (3)	0.001 (2)
C16	0.086 (4)	0.107 (5)	0.081 (4)	0.002 (3)	0.043 (3)	0.001 (3)
C17	0.098 (4)	0.107 (5)	0.123 (5)	-0.006 (4)	0.066 (4)	-0.008 (4)
C18	0.148 (6)	0.090 (5)	0.126 (6)	-0.017 (5)	0.095 (5)	0.000 (4)
C19	0.159 (6)	0.108 (5)	0.101 (5)	0.016 (5)	0.082 (5)	0.031 (4)

supplementary materials

C20	0.109 (4)	0.097 (4)	0.082 (4)	0.010 (4)	0.056 (3)	0.026 (3)
C21	0.064 (3)	0.038 (2)	0.043 (2)	-0.002 (2)	0.0134 (19)	0.0020 (18)
C22	0.080 (3)	0.059 (3)	0.047 (2)	-0.020 (2)	0.020 (2)	-0.006 (2)
C23	0.077 (3)	0.060 (3)	0.042 (2)	-0.003 (2)	0.017 (2)	0.003 (2)
C24	0.088 (4)	0.096 (4)	0.060 (3)	-0.034 (3)	0.027 (3)	-0.003 (3)
C25	0.133 (5)	0.148 (6)	0.070 (4)	-0.062 (5)	0.047 (4)	-0.002 (4)
C26	0.138 (5)	0.144 (6)	0.055 (3)	-0.030 (5)	0.042 (3)	0.008 (4)
C27	0.125 (5)	0.113 (5)	0.045 (3)	-0.022 (4)	0.015 (3)	-0.009 (3)
C28	0.095 (4)	0.081 (4)	0.053 (3)	-0.022 (3)	0.017 (2)	-0.010 (3)
C29	0.053 (2)	0.052 (3)	0.037 (2)	-0.004 (2)	0.0066 (18)	-0.0027 (18)
C30	0.066 (3)	0.047 (3)	0.061 (3)	0.003 (2)	0.007 (2)	-0.006 (2)
C31	0.148 (3)	0.098 (2)	0.182 (4)	0.017 (2)	0.089 (3)	0.031 (2)
C32	0.148 (3)	0.098 (2)	0.182 (4)	0.017 (2)	0.089 (3)	0.031 (2)
C33	0.148 (3)	0.098 (2)	0.182 (4)	0.017 (2)	0.089 (3)	0.031 (2)
C34	0.148 (3)	0.098 (2)	0.182 (4)	0.017 (2)	0.089 (3)	0.031 (2)
C35	0.148 (3)	0.098 (2)	0.182 (4)	0.017 (2)	0.089 (3)	0.031 (2)
C36	0.148 (3)	0.098 (2)	0.182 (4)	0.017 (2)	0.089 (3)	0.031 (2)

Geometric parameters (\AA , $^\circ$)

La1—O1	2.501 (3)	C11—C12	1.413 (6)
La1—O4	2.417 (3)	C11—H11	0.9300
La1—O7	2.564 (2)	C12—H12	0.9300
La1—O8	2.826 (3)	C13—C14	1.518 (5)
La1—N1	2.750 (3)	C14—H14A	0.9700
La1—N2	2.707 (3)	C14—H14B	0.9700
La1—O8 ⁱ	2.436 (3)	C15—C16	1.362 (7)
La1—O2 ⁱ	2.538 (2)	C15—C20	1.363 (7)
La1—O5 ⁱⁱ	2.551 (3)	C16—C17	1.392 (7)
O1—C13	1.255 (4)	C16—H16	0.9300
O2—C13	1.251 (4)	C17—C18	1.351 (8)
O2—La1 ⁱ	2.538 (2)	C17—H17	0.9300
O3—C15	1.375 (5)	C18—C19	1.349 (8)
O3—C14	1.422 (5)	C18—H18	0.9300
O4—C21	1.259 (4)	C19—C20	1.398 (7)
O5—C21	1.230 (5)	C19—H19	0.9300
O5—La1 ⁱⁱ	2.551 (3)	C20—H20	0.9300
O6—C23	1.376 (5)	C21—C22	1.517 (5)
O6—C22	1.417 (4)	C22—H22A	0.9700
O7—C29	1.241 (5)	C22—H22B	0.9700
O8—C29	1.268 (5)	C23—C24	1.365 (6)
O8—La1 ⁱ	2.436 (3)	C23—C28	1.383 (6)
O9—C31	1.343 (8)	C24—C25	1.390 (7)
O9—C30	1.408 (5)	C24—H24	0.9300
N1—C1	1.334 (5)	C25—C26	1.350 (8)
N1—C5	1.356 (5)	C25—H25	0.9300
N2—C12	1.327 (5)	C26—C27	1.362 (8)
N2—C9	1.354 (5)	C26—H26	0.9300

C1—C2	1.386 (6)	C27—C28	1.383 (7)
C1—H1	0.9300	C27—H27	0.9300
C2—C3	1.347 (8)	C28—H28	0.9300
C2—H2	0.9300	C29—C30	1.509 (5)
C3—C4	1.396 (8)	C30—H30A	0.9700
C3—H3	0.9300	C30—H30B	0.9700
C4—C5	1.418 (6)	C31—C32	1.387 (11)
C4—C6	1.457 (8)	C31—C36	1.392 (11)
C5—C9	1.436 (6)	C32—C33	1.425 (9)
C6—C7	1.328 (9)	C32—H32	0.9300
C6—H6	0.9300	C33—C34	1.347 (10)
C7—C8	1.418 (8)	C33—H33	0.9300
C7—H7	0.9300	C34—C35	1.402 (11)
C8—C10	1.390 (8)	C34—H34	0.9300
C8—C9	1.420 (6)	C35—C36	1.383 (9)
C10—C11	1.353 (7)	C35—H35	0.9300
C10—H10	0.9300	C36—H36	0.9300
O1—La1—O4	145.89 (9)	C12—C11—H11	120.7
O1—La1—O7	73.58 (9)	N2—C12—C11	122.1 (5)
O1—La1—O8	65.02 (8)	N2—C12—H12	118.9
O4—La1—O7	138.87 (9)	C11—C12—H12	118.9
O4—La1—O8	139.82 (8)	O2—C13—O1	127.7 (4)
O7—La1—O8	47.79 (7)	O2—C13—C14	119.2 (4)
O1—La1—N1	127.85 (9)	O1—C13—C14	113.1 (3)
O4—La1—N1	76.73 (9)	O3—C14—C13	111.1 (3)
O7—La1—N1	63.87 (9)	O3—C14—H14A	109.4
O8—La1—N1	102.38 (9)	C13—C14—H14A	109.4
O1—La1—N2	81.11 (9)	O3—C14—H14B	109.4
O4—La1—N2	96.46 (9)	C13—C14—H14B	109.4
O7—La1—N2	74.58 (9)	H14A—C14—H14B	108.0
O8—La1—N2	118.27 (8)	C16—C15—C20	120.2 (5)
N1—La1—N2	60.16 (11)	C16—C15—O3	114.8 (4)
O4—La1—O8 ⁱ	88.28 (10)	C20—C15—O3	125.0 (4)
O8 ⁱ —La1—O1	77.78 (9)	C15—C16—C17	119.9 (6)
O4—La1—O2 ⁱ	75.76 (9)	C15—C16—H16	120.0
O8 ⁱ —La1—O2 ⁱ	74.99 (9)	C17—C16—H16	120.0
O1—La1—O2 ⁱ	128.31 (9)	C18—C17—C16	120.1 (6)
O4—La1—O5 ⁱⁱ	77.28 (9)	C18—C17—H17	120.0
O8 ⁱ —La1—O5 ⁱⁱ	78.07 (10)	C16—C17—H17	120.0
O1—La1—O5 ⁱⁱ	69.48 (9)	C19—C18—C17	120.0 (6)
O2 ⁱ —La1—O5 ⁱⁱ	142.10 (9)	C19—C18—H18	120.0
O8 ⁱ —La1—O7	120.43 (9)	C17—C18—H18	120.0
O2 ⁱ —La1—O7	83.70 (9)	C18—C19—C20	120.9 (6)
O5 ⁱⁱ —La1—O7	133.46 (8)	C18—C19—H19	119.5
O8 ⁱ —La1—N2	148.44 (10)	C20—C19—H19	119.5
O2 ⁱ —La1—N2	136.43 (10)	C15—C20—C19	118.9 (6)

supplementary materials

O5 ⁱⁱ —La1—N2	72.66 (10)	C15—C20—H20	120.6
O8 ⁱ —La1—N1	150.22 (10)	C19—C20—H20	120.6
O2 ⁱ —La1—N1	76.43 (9)	O5—C21—O4	127.2 (4)
O5 ⁱⁱ —La1—N1	122.08 (10)	O5—C21—C22	119.7 (4)
O8 ⁱ —La1—O8	72.87 (10)	O4—C21—C22	113.0 (4)
O2 ⁱ —La1—O8	65.31 (8)	O6—C22—C21	111.5 (3)
O5 ⁱⁱ —La1—O8	129.85 (9)	O6—C22—H22A	109.3
C13—O1—La1	130.5 (2)	C21—C22—H22A	109.3
C13—O2—La1 ⁱ	137.6 (3)	O6—C22—H22B	109.3
C15—O3—C14	117.3 (3)	C21—C22—H22B	109.3
C21—O4—La1	151.7 (3)	H22A—C22—H22B	108.0
C21—O5—La1 ⁱⁱ	149.8 (3)	C24—C23—O6	124.8 (4)
C23—O6—C22	117.6 (3)	C24—C23—C28	119.9 (4)
C29—O7—La1	101.5 (2)	O6—C23—C28	115.2 (4)
C29—O8—La1 ⁱ	162.7 (3)	C23—C24—C25	119.2 (5)
C29—O8—La1	88.3 (2)	C23—C24—H24	120.4
La1 ⁱ —O8—La1	107.13 (10)	C25—C24—H24	120.4
C31—O9—C30	118.1 (5)	C26—C25—C24	121.2 (5)
C1—N1—C5	117.8 (4)	C26—C25—H25	119.4
C1—N1—La1	120.9 (3)	C24—C25—H25	119.4
C5—N1—La1	119.8 (3)	C25—C26—C27	119.6 (5)
C12—N2—C9	119.0 (4)	C25—C26—H26	120.2
C12—N2—La1	119.5 (3)	C27—C26—H26	120.2
C9—N2—La1	121.3 (3)	C26—C27—C28	120.7 (5)
N1—C1—C2	123.3 (5)	C26—C27—H27	119.6
N1—C1—H1	118.4	C28—C27—H27	119.6
C2—C1—H1	118.4	C23—C28—C27	119.3 (5)
C3—C2—C1	119.4 (5)	C23—C28—H28	120.4
C3—C2—H2	120.3	C27—C28—H28	120.4
C1—C2—H2	120.3	O7—C29—O8	122.2 (4)
C2—C3—C4	120.3 (5)	O7—C29—C30	120.5 (4)
C2—C3—H3	119.9	O8—C29—C30	117.3 (4)
C4—C3—H3	119.9	O9—C30—C29	113.8 (3)
C3—C4—C5	117.3 (5)	O9—C30—H30A	108.8
C3—C4—C6	124.5 (6)	C29—C30—H30A	108.8
C5—C4—C6	118.2 (6)	O9—C30—H30B	108.8
N1—C5—C4	122.0 (5)	C29—C30—H30B	108.8
N1—C5—C9	118.2 (4)	H30A—C30—H30B	107.7
C4—C5—C9	119.8 (4)	O9—C31—C32	123.7 (8)
C7—C6—C4	121.5 (6)	O9—C31—C36	114.0 (9)
C7—C6—H6	119.3	C32—C31—C36	122.3 (8)
C4—C6—H6	119.3	C31—C32—C33	117.9 (9)
C6—C7—C8	121.6 (5)	C31—C32—H32	121.1
C6—C7—H7	119.2	C33—C32—H32	121.1
C8—C7—H7	119.2	C34—C33—C32	118.4 (9)
C10—C8—C7	123.3 (5)	C34—C33—H33	120.8
C10—C8—C9	117.0 (5)	C32—C33—H33	120.8

C7—C8—C9	119.7 (5)	C33—C34—C35	124.4 (8)
N2—C9—C8	122.0 (4)	C33—C34—H34	117.8
N2—C9—C5	119.0 (3)	C35—C34—H34	117.8
C8—C9—C5	119.0 (5)	C36—C35—C34	117.0 (9)
C11—C10—C8	121.1 (5)	C36—C35—H35	121.5
C11—C10—H10	119.4	C34—C35—H35	121.5
C8—C10—H10	119.4	C35—C36—C31	119.7 (9)
C10—C11—C12	118.7 (5)	C35—C36—H36	120.1
C10—C11—H11	120.7	C31—C36—H36	120.1

Symmetry codes: (i) $-x+2, -y, -z+2$; (ii) $-x+2, -y+1, -z+2$.

Hydrogen-bond geometry ($\text{\AA}, ^\circ$)

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C10—H10 \cdots O7 ⁱⁱⁱ	0.93	2.33	3.196 (5)	156
C30—H30B \cdots O4 ^{iv}	0.97	2.39	3.339 (5)	166
C12—H12 \cdots O5 ⁱⁱ	0.93	2.45	3.039 (5)	122
C1—H1 \cdots O2 ⁱ	0.93	2.45	3.132 (5)	130

Symmetry codes: (iii) $-x+3/2, y+1/2, -z+3/2$; (iv) $x, y-1, z$; (ii) $-x+2, -y+1, -z+2$; (i) $-x+2, -y, -z+2$.

Article retracted

Fig. 1

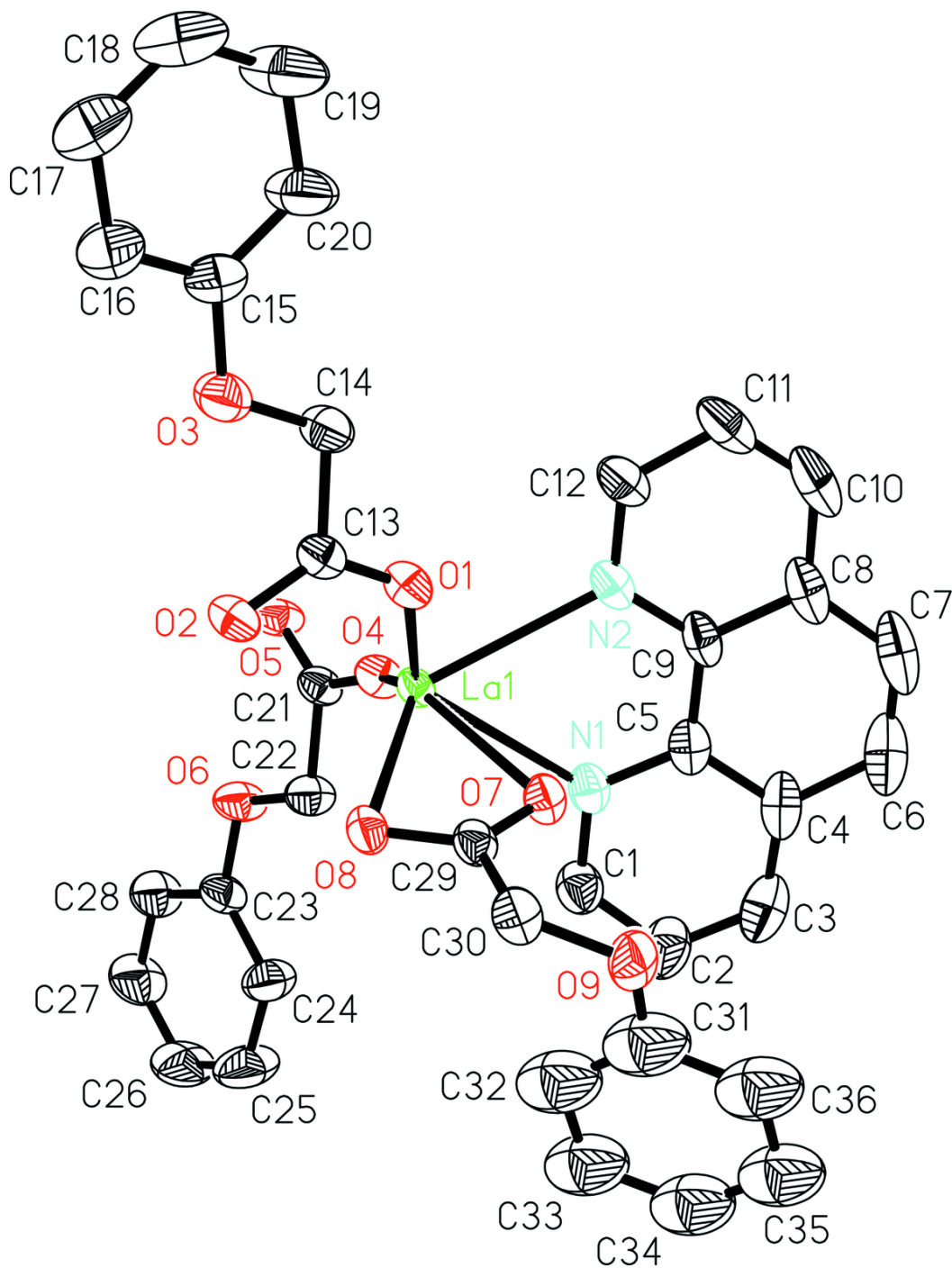
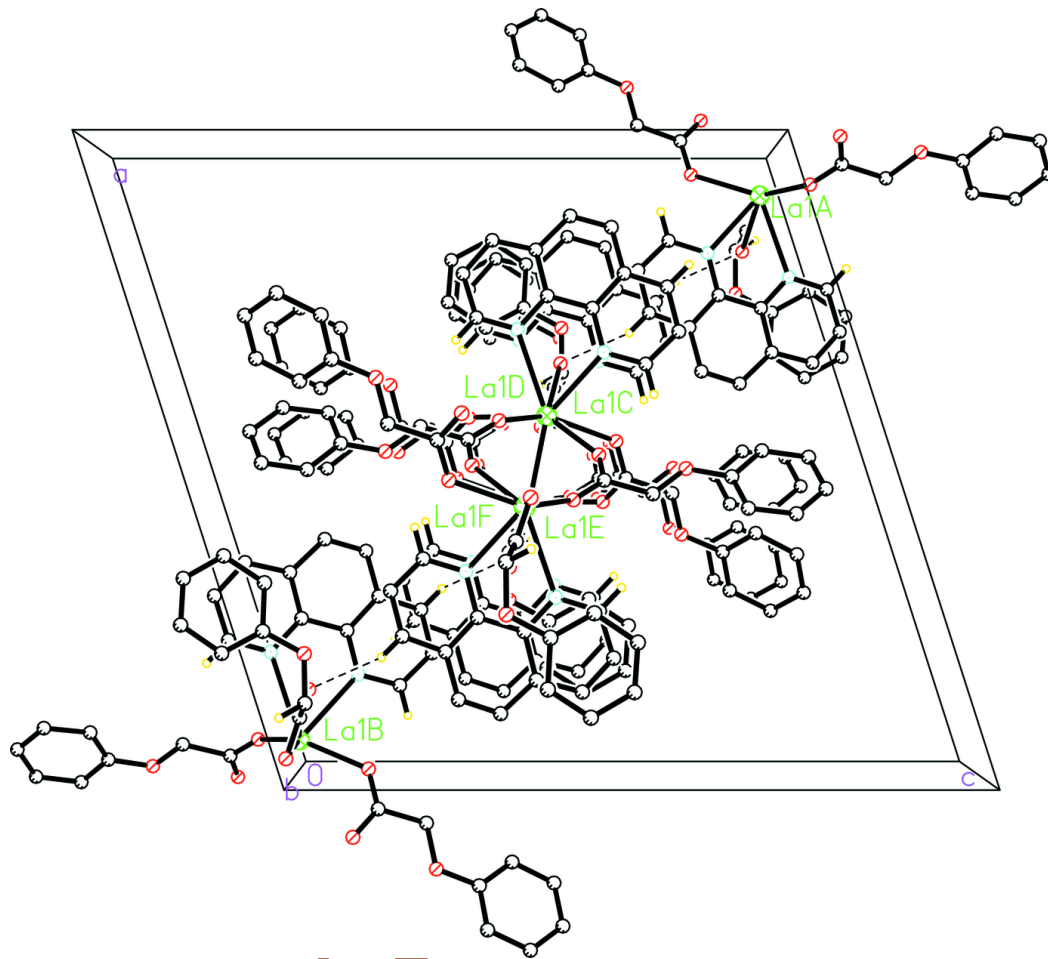


Fig. 2



Article